

Department of Chemistry and Chemical Sciences CENTRAL UNIVERSITY OF JAMMU Rahya-Suchani (Bagla), District-Samba, Jammu-181143, (J&K) India

Five-year Integrated B.Sc. (Hons.) M.Sc. Chemistry Teaching Plan (Aug 2022-Dec 2022)

| Semester: | V |
|-----------------|---|
| Course: | Properties of Inorganic Metal Complexes (ICCHM4C005T) |
| Course Teacher: | Dr. Sujata Kundan |

| Week | Lecture No./Day | Topic to be Taught | No of Hours | Suggested Readings |
|-------------------------|--------------------|--|----------------|-----------------------|
| 1 st Week | Ι | Unit-I | 1 | 1, 2, 3, 9 |
| | | Electronic spectra of transition metal complexes-I: | | |
| | | Quantum numbers, Types of electronic transitions. | | |
| | II | Selection rules for <i>d</i> - <i>d</i> transitions, Spectroscopic ground | 1 | 1, 2, 3, 9 |
| | | states. | | |
| | III | Term symbols, Microstates, Spectrochemical series of | 1 | 1, 2, 3, 9 |
| | | ligands. | | |
| | IV | Orbital and spin magnetic moments, Orbital contribution, | 1 | 1, 2, 3, 9 |
| l | | Quenching of magnetic moment. | | |
| · | Ι | Russel-Saunders Coupling: <i>l-l</i> coupling, <i>J-J</i> coupling, <i>L-S</i> | 1 | 1, 2, 3, 9 |
| and | | coupling. | | |
| 2^{nd} | II | To be Contd | 1 | 1, 2, 3, 9 |
| Week | III | Derivation of Russell-Saunders terms: p^2 configuration. | 1 | 1, 2, 3, 9 |
| | IV | Derivation of Russell-Saunders terms: d^2 configuration. | 1 | 1, 2, 3, 9 |
| | Ι | Orgel energy level diagram for d^1 to d^9 states, | 1 | 1, 2, 3, 9 |
| | _ | Nepheleuxatic effect. | _ | -, _, _, , , |
| | II | To be Contd | 1 | 1, 2, 3, 9 |
| ord | III | To be Contd | 1 | 1, 2, 3, 9 |
| 3 rd | IV | Unit-II | 1 | 3, 4, 5, 6, 9 |
| Week | | Magnetic properties of transition metal complexes-I: | | - 7 7 - 7 - 7 - |
| | | Definition of magnetic properties, Types of magnetic | | |
| | | bodies: Diamagnetism, Paramagnetism Ferromagnetism, | | |
| | | Ferrimagnetism and Antiferromagnetism. | | |
| | Ι | To be Contd | 1 | 3, 4, 5, 6, 9 |
| Ath | II | Mechanism of anti-ferromagnetic interaction, Spin-only | 1 | 3, 4, 5, 6, 9 |
| 4 th | | formula. | | |
| Week | III | Spin orbit coupling, Lande interval rule. | 1 | 3, 4, 5, 6, 9 |
| | IV | Energies of <i>J</i> -levels. | 1 | 3, 4, 5, 6, 9 |
| | Ι | Curie equation, Curie and Curie-Weiss law. | 1 | 3, 4, 5, 6, 9 |
| | II | Temperature independent paramagnetism(TIP), Derivation | 1 | 3, 4, 5, 6, 9 |
| 5^{th} | | and application of Van Vleck susceptibility equation. | | |
| Week | III | Magnetic exchange coupling and spin crossover (Low spin | 1 | 3, 4, 5, 6, 9 |
| | | and high spin cross over). | | |
| | IV | Anomalous magnetic moments. | 1 | 3, 4, 5, 6, 9 |
| 6 th Week | Ι | Magnetic properties of binuclear and polynuclear | 1 | 3, 4, 5, 6, 9 |
| | | complexes. | | |
| | II | Unit-III | 1 | 3, 4, 5, 6, 9 |
| | | Magnetic properties of transition metal complexes-II: | | |
| | | Magnetic susceptibility-orbital and spin effects. | | |
| | III | Importance of magnetic susceptibility, Diamagnetism and | 1 | 3, 4, 5, 6, 9 |
| | | Pascals's constant. | | |



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| | IV | Gouy's method, Faraday method. | 1 | 3, 4, 5, 6, 9 |
|------------------|-----------|--|---|--------------------------------|
| | Ι | Vibrating sample magnetometer, SQUID, NMR method | 1 | 3, 4, 5, 6, 9 3, 4, 5, 6, 9 |
| | | for measuring magnetic susceptibility. | | - , , - , - , - |
| 7 th | II | To be Contd | 1 | 3, 4, 5, 6, 9 |
| | III | Correlation of μ s and μ eff values, Orbital contribution to | 1 | 3, 4, 5, 6, 9 3, 4, 5, 6, 9 |
| Week | | magnetic moments. | - | 0, 1, 0, 0, 2 |
| | IV | Magnetic properties based on crystal field models: | 1 | 3, 4, 5, 6, 9 |
| | | Octahedral, Tetrahedral. | - | 0, 1, 0, 0, , |
| | Ι | To be Contd | 1 | 3, 4, 5, 6, 9 |
| oth | II | Trigonal bipyramidal, Square pyramidal. | 1 | 3, 4, 5, 6, 9 |
| 8 th | III | To be Contd | 1 | 3, 4, 5, 6, 9 |
| Week | IV | tetragonally distorted octahedral complexes, Diamagnetism | 1 | 3, 4, 5, 6, 9 |
| | | in atoms and polynuclear systems. | | , , , , , |
| | Ι | To be Contd | 1 | 3, 4, 5, 6, 9 |
| Γ | II | Unit-IV | 1 | 1, 2, 7, 8, 9 |
| | | Metal π -complexes–I: Metal carbonyls, Classification of | | |
| 9 th | | metal carbonyls. | | |
| Week | III | Effective atomic number, Preparation and important | 1 | 1, 2, 7, 8, 9 |
| WEEK | | reactions (substitution, nucleophilic, electrophilic, | | |
| | | reduction reactions) of metal carbonyls, Structure and | | |
| | | chemical bonding in metal carbonyls. | | |
| | IV | To be Contd | 1 | 1, 2, 7, 8, 9 |
| | Ι | To be Contd | 1 | 1, 2, 7, 8, 9 |
| 10 th | II | To be Contd | 1 | 1, 2, 7, 8, 9 |
| Week | III | Preparation of anionic metal carbonyl complexes and | 1 | 1, 2, 7, 8, 9 |
| WEEK | | Substituted metal carbonyl complexes. | | |
| | IV | To be Contd | 1 | 1, 2, 7, 8, 9 |
| | Ι | Vibrational spectra of metal carbonyls for bonding and | 1 | 1, 2, 7, 8, 9 |
| | | structural elucidation, Application of metal carbonyls | | |
| _ | | complexes. | | |
| 11 th | II | To be Contd | 1 | 1, 2, 7, 8, 9 |
| Week | III | To be Contd | 1 | 1, 2, 7, 8, 9 1, 2, 7, 8, 9 |
| | IV | Unit-V | 1 | 1, 2, 7, 8, 9 |
| | | Metal π -complexes-II: Dinitrogen complexes- | | |
| | | Preparation, Structure, Bonding, and important reactions | | |
| | T | with transition metals. | 1 | 10700 |
| _ | I | To be Contd | 1 | 1, 2, 7, 8, 9 |
| 12 th | II | To be Contd | 1 | 1, 2, 7, 8, 9 |
| Week | III | Dioxygen complexes-Preparation, Structure, Bonding, and | 1 | 1, 2, 7, 8, 9 |
| _ | IV | important reactions with transition metals. | 1 | 1 2 7 8 0 |
| | | To be Contd | 1 | 1, 2, 7, 8, 9 |
| F | I II | To be Contd | 1 | 1, 2, 7, 8, 9 |
| 13^{th} | II | Metal nitrosyls complexes-Preparation, Structure, | 1 | 1, 2, 7, 8, 9 |
| Week | III | Bonding, and important reactions with transition metals.To be Contd | 1 | 12780 |
| | III IV | To be Contd | 1 | 1, 2, 7, 8, 9 |
| | | | | 1, 2, 7, 8, 9 |
| 14 th | I II | Ligating behaviour of tertiary phosphines. | 1 | 1, 2, 7, 8, 9 |
| Week | | Isopoly and heteropoly acids. | 1 | 1, 2, 7, 8, 9 |
| | III IV | To be Contd To be Contd | 1 | 1, 2, 7, 8, 9 |
| | 1 V | | 1 | 1, 2, 7, 8, 9 |



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| 15 th Week | Ι | Salts of molybdenum and tungsten. | 1 | 1, 2, 7, 8, 9 |
|--------------------------|-----|--|---|---------------|
| | Π | To be Contd | 1 | 1, 2, 7, 8, 9 |
| | III | To be Contd | 1 | 1, 2, 7, 8, 9 |
| | IV | Revision of Units I, Discussion of model question papers | 1 | 1,7 |
| 16 th Week | Ι | Revision of Units II, Discussion of model question papers | 1 | |
| | Π | Revision of Units III, Discussion of model question papers | 1 | |
| | III | Revision of Unit IV, Discussion of model question papers | 1 | |
| | IV | Revision of Units V, Discussion of model question papers | 1 | |

REFERENCES

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- 2. Malik, Tuli and Madan, *Selected Topics in Inorganic Chemistry*, S. Chand & company, New Delhi, 2009.
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- 4. A. Earnshaw, Introduction to Magnetochemistry, 1st Ed., Academic Press, 1968.
- 5. F. B. Mabbs, D. J. Machin, et al., *Magnetism and Transition Metal Complexes*, Dover Books, 2008.
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- 8. J. D. Lee, Concise Inorganic Chemistry, 5th Ed., John Wiley & Sons, 2008.
- 9. In-house Study material.